

Report of the third interlaboratory comparison organised by the Community Reference Laboratory for Heavy Metals in Feed and Food

Total Cd, Pb and Hg and extractable Cd and Pb in feed

M.B. de la Calle, J. van de Kreeke, I. Verbist, S. Bynens, P. Taylor



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European Commission Joint Research Centre Institute for Reference Materials and Measurements

Contact information

Address: European Commission, Joint Research Centre, Institute for Reference Materials and Measurements, Retieseweg 111, 2440 Geel, Belgium E-mail: Maria.de-la-calle@ec.europa.eu Tel.: +32 (0)14 571252 Fax: +32 (0)14 571863

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Heavy Metals in Feed and Food

Report of the third interlaboratory comparison Total Cd, Pb and Hg and extractable Cd and Pb in feed



January 2007

María Beatriz de la Calle Johannes van de Kreeke Inge Verbist Saskia Bynens Philip Taylor

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1 Summary

The Institute for Reference Materials and Measurements (IRMM) of the Joint Research Centre, a Directorate-General of the European Commission, operates the Community Reference Laboratory for Heavy Metals in Feed and Food (CRL-HM). One of its core tasks is to organise interlaboratory comparisons (ILCs) among appointed National Reference Laboratories (NRLs). This report presents the results of the third ILC of the CRL-HM which focused on the determination of total Cd, Pb and Hg and extractable Cd and Pb in feed according to Directive $2002/32/EC^1$ of the European Parliament and of the Council on undesirable substances in animal feed.

The test material used in this exercise was a commercial compound feed for fish provided by the Centro di Referenza Nazionale per la Sorveglianza e il Controllo degli Alimenti per Animali, Istituto Zooprofilattico del Piemonte. The material, naturally contaminated, was processed, bottled, labelled and dispatched by the Reference Materials Unit of the IRMM. The samples were dispatched on the second half of October 2007. Each participant received one bottle containing approximately 20 g of test material. Thirty-one participants from 25 countries registered to the exercise of which 31 submitted results for total Cd and total Pb, 28 for total Hg, 26 for extractable Cd and 24 for extractable Pb.

The assigned values (X_{ref}) were provided by IRMM using isotope dilution-inductively coupled plasma-mass spectrometry (ID-ICP-MS). The analytical uncertainty of X_{ref} , u_{char} , was calculated according to the ISO Guide to the Expression of Uncertainty in Measurement (GUM)². Homogeneity and stability studies were subcontracted to Bundesanstalt für Materialforschung und –Prüfung (BAM). The uncertainties of the respective assigned values, u_{ref} , were calculated combining the analytical uncertainty, u_{char} , with a contribution for the between-bottle homogeneity, u_{bb} , and for the short term stability of the test material, u_{sts} . Participants were invited to report the uncertainty on their measurements. This was done by 29 laboratories for total Cd, 23 for total Pb, 25 for total Hg, 18 for extractable Cd and 13 for extractable Pb.

Laboratory results were rated with z and zeta scores in accordance with ISO 13528^3 . Standard deviation for proficiency assessment (also called target standard deviation) were calculated using the modified Horwitz equation⁴ and were 22 % of the assigned value for total Pb and Hg and for extractable Pb and 16 % of the assigned value for total and extractable Cd.

2 Introduction

In humans, about one third of the total cadmium burden originates from animal products. As a result of contaminated animal feed, cadmium is ubiquitous in food of animal origin. The rate of oral uptake in animals exceeds excretion and so accumulation results in some animal tissues, mainly in kidneys and liver. Cadmium-containing dust can increase the concentration of cadmium in plants by resorption via the leaves. The cadmium contents of vegetable feeds vary from 0.005 to 0.1 mg kg⁻¹, and in certain cases up to 1 mg kg⁻¹. Very likely cadmium accumulates in marine organisms⁵. Fresh water fish and shellfish of polluted areas show cadmium levels of about 0.03 mg kg⁻¹. Accumulation of cadmium in animals can be avoided if the feed concentration does not exceed 0.6 mg kg⁻¹, and drinking water for animals should not contain more than 0.05 mg L^{-1 6}.

About half of human lead intake is through food, of which more than half originates from plants. Lead found in food and animal feed comes mainly from external sources. Plant feed have lead contents ranging from less than 0.1 mg kg⁻¹ up to 5 to 10 mg kg⁻¹ of dry matter. The lead content of animal tissues is caused mainly by the uptake of contaminated feed, only 10 % of the total burden is caused by the inhalation of lead containing dust. To avoid accumulation, animal drinking water should not contain more than 0.1 mg kg⁻¹. Lead does not play a major role in aquatic food chains. Lead levels in fish depend upon the amount of lead pollution in the environmental water and there is remarkable variation in the existing data in particular among different species of fish. In mammals the amount of lead in the organism can be divided into three fractions: blood lead and some rapidly exchanging soft tissues with a half-life of about 19 days, soft tissues and a rapidly exchangeable bone fraction with a half life of about 21 days and bones and the skeleton with a half life of about 10 to 20 years. Similarly, Alves and Wood⁷ indicate that bone accumulates the most lead per fish weight in some fish species (rainbow trout). Nevertheless, not a lot of information exists in the literature about the metabolism of lead in fish.

Food and animal feed derived from plants usually have mercury contents between 0.001 and 0.03 mg kg⁻¹. Through industrial usage, mercury is sometimes introduced into bodies of water. Marine organisms are especially able to transform inorganic species of mercury into organic compounds, which make mercury easily transferred through the aquatic food chain. Accumulation factors in the marine food chain are 100 to 1000 compared to 2 to5 in the terrestrial food chain. Thus, seafood is a particular source of mercury burden in man and animals. Mercury passes to domestic animals and animal-derived food products via fish meal which is in particular widely used in the feed for poultry.

To overcome problems associated with a high metal content in feed maximum levels in several commodities have been laid down in Directive 2002/32/EC, and a network has been built up to ensure quality and comparability in official controls throughout the European Union.⁸ In March 2006 a footnote was introduced in Directive 2002/32/EC in which it is stated that *"Maximum levels refer to an analytical determination of lead and cadmium, whereby extraction is performed in nitric acid (5 % w/w) for 30 minutes at boiling temperature"*.

The Community Reference Laboratory for Heavy Metals in Feed and Food has organised a proficiency test (PT) exercise for the network of appointed National Reference Laboratories (NRLs) to determine the total Cd, Pb and Hg and extractable Cd and Hg. The two later measurands are to be determined using an extraction procedure agreed upon by the CRL-HM and the network of NRLs, and which is in agreement with the requirements laid down in Directive 2002/32/EC.

3 Scope

As stated in Regulation 882/2004 of the European Parliament and of the Council⁸, one of the core duties of the CRL-HM is to organise interlaboratory comparisons for the benefit of staff from National Reference Laboratories. The scope of this ILC is to test the competence of the appointed NRLs to determine the total concentration of Cd, Pb and Hg and of extractable Cd and Pb according to Directive 2002/32/EC.

The assessment of the measurement results is undertaken on the basis of requirements laid down in legislation¹, and follows the administrative and logistic procedures of IMEP⁹, the International Measurement Evaluation Programme of the IRMM of the European Commission's Directorate Joint Research Centre. The designation of this ILC is IMEP-103.

4 Time frame

This interlaboratory comparison was agreed upon by the NRLs network at the first CRL-HM workshop held on 25/26 September 2006. Specific details of the exercise were refined during the second CRL-HM workshop held on 24/25 September 2007. Invitation letters were sent to the participants on 5 October 2007. (cf. Annex 1). The samples were dispatched to the participants on 18 October 2007. Reporting deadline was 30 November 2007.

5 Test material

5.1 Preparation

The test material, commercially available feed for fish, was provided by the Centro di Referenza Nazionale per la Sorveglianza e il Controllo degli Alimenti per Animali, Istituto Zooprofilattico del Piemonte. Upon arrival at IRMM the material was processed by the Reference Materials Unit in the following way: the particle size distribution was assessed by laser diffraction and the water content determined by Karl-Fisher titration. Coarse particles were removed sieving through a 500 μ m sieve and the water content was reduced to about 5 % by vacuum drying. The material was then homogenised and distributed (vibrating feeder) into amber glass bottles with polyethylene (PE) insert and screw cap lid with crimp film, containing approximately 20 g of test material each. Before and after processing, the material was stored at -20 °C. The processing took place at room temperature.

5.2 Homogeneity and stability

The measurements for homogeneity and stability studies were performed by Bundesanstalt für Materialforschung und -prüfung (BAM). Homogeneity was evaluated according to the method proposed by Fearn and Thompson¹⁰ (one of the approaches recommended by the IUPAC International Harmonised Protocol¹¹) and to the method proposed in the ISO 13528. The material proved to be homogeneous according to the IUPAC International Harmonised Protocol for the five measurands. The test material was also homogeneous for the total content of Hg according to ISO 13528. For the total and extractable Cd the test material was not homogeneous according to ISO13528. Unfortunately, the homogeneity measurements for both total and extractable Cd were not performed in a random way but following the filling sequence, which made it not possible to determine if the observed drift between bottles was due to the filling or to the analytical process. A contribution to the assigned uncertainty, u_{ref}, due to likely heterogeneity of the material, u_{bb}, of 6 % as provided by the Soft CRM software¹² licensed to the Reference Material Unit of IRMM, was taken on board following ISO Guide 35¹³. For total and extractable lead, the material was not homogeneous according to ISO 13528, and in fact even high within-bottle heterogeneity was found. A similar problem has been previously observed at IRMM (data not published) with a fish material. Intrinsic micro-heterogeneity of the material due to the fact that most likely the largest fraction of lead in fish is found in bones, as indicated in the introduction⁷, could be an explanation for the obtained results.

The study of the stability of the test material was conducted following the isochronous approach¹⁴. The evaluation of the stability of the test material was made using the Soft CRM software¹⁵. The material proved to be stable at 4 °C for the six weeks that elapsed between the dispatch of the samples and the deadline for submission of results for total Cd and Hg and for extractable Cd. No sound stability statement could be made for total and extractable Pb for the reasons already explained above.

The analytical results and statistical evaluation of the homogeneity and short term stability studies are provided in Annex 2.

5.3 Distribution

One set of material was sent to every participant. The test material was dispatched to the participants by IRMM on 18 October 2007. Each participant received: a) one bottle containing approximately 20 g of test material (one laboratory received two bottles because the method to be used, dry ashing, required the use for 5 g of material per replicate and 20 g would not be enough for the five measurands plus the water content determination), b) and accompanying letter with instructions on sample handling and reporting and with the method to be applied for the determination of extractable Cd and Pb (cf. Annex 3) and c) a form which had to be sent back after receipt of the sample to confirm its arrival (cf. Annex 4).

6 Instructions to participants

Details on this ILC were discussed with the NRLs at the first workshop. Concrete instructions were given to all participants in a letter that accompanied the samples (Annex 3). The measurands and matrix were clearly defined as "<u>Total</u> Cd, Pb and Hg and <u>extractable</u> <u>amounts</u> of Cd and Pb in feed following Directive 2002/32/EC of the European Parliament and of the Council on undesirable substances in animal feed".

Laboratories were asked to perform two or three independent measurements and report them, together with the mean of the results and its associated uncertainty. Some laboratories reported four independent results. Participants were asked to follow their routine procedures for the determination of total Cd, Pb and Hg and the procedure previously agreed upon for the determination of extractable Cd and Pb. The results were to be reported in the same manner (eg. number of significant figures) as when reporting to customers.

The results were to be reported in a special on line form for which every participant received an individual access code. A special questionnaire, aiming to collect additional information, was included in the online form. The questionnaire is presented in Annex 5.

7 Reference values and their standard uncertainties

The reference value used, X_{ref} , for this ILC was determined by IRMM using Isotope Dilution Inductively Coupled Plasma Mass Spectrometry (ID-ICP-MS). IRMM has proven its measurement capabilities by successful participation in CCQM key comparisons. No reference value was provided for total and extractable Pb due to the heterogeneity problems also found by BAM. The standard uncertainty associated to the assigned value (u_{ref}) was calculated as:

$$u_{\text{ref}} = \sqrt{u_{\text{char}}^2 + u_{\text{bb}}^2 + u_{\text{lts}}^2}$$
 Eq. 1

where:

u_{ref} standard uncertainty associated to the assigned value

u_{char} standard uncertainty of characterisation

u_{bb} contribution for the between-bottle homogeneity

u_{sts} standard uncertainty contribution derived from the short-term-stability study

The values of X_{ref} , u_{char} , u_{bb} , u_{sts} , u_{ref} and the expanded standard uncertainty U_{ref} , are summarised in Table 1.

		u _{char} [%]	u_{bb} [%]	u _{sts} * [%]	u _{ref} [%]	
Total Pb			10	20.9		
Extract. Pb			12	23.9		
Total Cd	1.100	1.55	6	2.30	6.61	0.145
Extract. Cd	1.100	1.55	6	1.40	6.61	0.145
Total Hg	0.0499	3.51	1	0.60	3.70	0.004

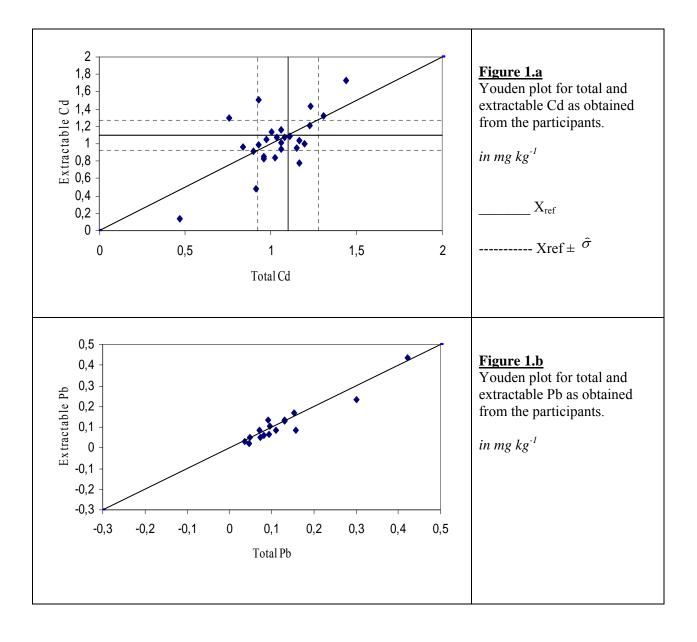
Table 1: assigned values and their standard uncertainties for the parameters of this ILC.

* For six weeks,

 X_{ref} is the certified reference value and u_{ref} the corresponding standard uncertainty; U_{ref} is the estimated expanded uncertainty, with a coverage factor k=2, corresponding to a level of confidence of about 95%.

As summarised in Table 1, total digestion and partial extraction of the test material, following the procedure described in the accompanying letter to the participants, provide identical Cd concentrations. This finding is supported by the Youden plot, Figure 1.a, constructed with the results provided by the participants in this exercise. One single cloud of points is observed on both axes around the reference value when total vs extractable Cd is plotted, showing that the results are not dependent of the method applied.

Unfortunately, due to the heterogeneity problems already described no reference value was provided for Pb, total or extractable. The Youden plot is also shown for lead, Figure 1.b, with no indication of method dependence. Nevertheless, more studies are needed before extracting any definitive conclusion on the Pb matter.



8 Evaluation of results

8.1 General observations

Thirty-one laboratories from 25 countries registered for participation in this exercise. Thirtyone laboratories submitted results for total Cd and total Pb (5 laboratories reported "< than" for total Pb), 28 for total Hg (two out of the 28 reported "< than "), 26 for extractable Cd and 24 for extractable Pb (7 laboratories out of the 24 reported "< than"). All laboratories reported two or more measurement values. Twenty-nine laboratories reported uncertainty for total Cd, 23 for total Pb, 25 for total Hg, 18 for extractable Cd and 13 for extractable Pb. All laboratories responded to the questionnaire included in the online reporting form.

8.2 Scores and evaluation criteria

Individual laboratory performance is expressed in terms of z and zeta scores in accordance with ISO 13528^3 and the International Harmonised Protocol¹¹.

$$z = \frac{x_{lab} - X_{ref}}{\hat{\sigma}}$$
 Eq. 2

$$zeta = \frac{x_{lab} - X_{ref}}{\sqrt{u^2_{ref} + u^2_{lab}}}$$
Eq. 3

where:

 $\begin{array}{ll} x_{lab} & \text{is the measurement result reported by a participant} \\ X_{ref} & \text{is the certified reference value (assigned value)} \\ u_{ref} & \text{is the standard uncertainty of the reference value} \\ u_{lab} & \text{is the standard uncertainty reported by a participant} \\ \hat{\sigma} & \text{is the standard deviation for proficiency assessment} \end{array}$

The z score compares the participant's deviation from the reference value with the standard deviation accepted for the proficiency test, $\hat{\sigma}$. Very frequently, in the area of food and feed $\hat{\sigma}$ is derived from the improved Horwitz equation⁴. The values for $\hat{\sigma}$ obtained for this exercise when applying the improved Horwitz equation were 22 % for total Hg and 16% for total and extractable Cd.

No scoring is provided for total and extractable Pb, in view of the heterogeneity problems. This fact made it not possible even for internationally recognised expert laboratories in the field, such as BAM and IRMM, to provide sound values for total and extractable Pb.

Should participants feel that these $\hat{\sigma}$ values are not fit for their purpose they can recalculate their scorings with a standard deviation matching their requirements. x_{lab} is the mean of the individual measurement results calculated by the ILC organiser. If reproducibility of 22 % for

total mercury and 16 % for total and extractable Cd are regarded as satisfactory, the z-score can be interpreted as:

$$\begin{split} |z| &\leq 2 \quad \text{satisfactory result} \\ 2 &< |z| &\leq 3 \quad \text{questionable result} \\ |z| &> 3 \quad \text{unsatisfactory result} \end{split}$$

Zeta score state if the laboratory result agree with the assigned value within the respective uncertainties. The interpretation of the zeta score is similar to the interpretation of the z-score:

$$\begin{split} |z| &\leq 2 \quad \text{satisfactory result} \\ 2 &< |z| &\leq 3 \quad \text{questionable result} \\ |z| &> 3 \quad \text{unsatisfactory result} \end{split}$$

An unsatisfactory zeta-score might be due to an underestimation of the uncertainty, or to a large error causing a large deviation from the reference value, or to a combination of the two factors. A laboratory with an unsatisfactory zeta-score has an estimation of the uncertainty of its measurements which is not consistent with the laboratory's deviation from the reference value. Zeta-scores were only provided for total Cd and total Hg. It was understood by the organiser that laboratories have had little time for the implementation of the method for the partial extraction of Cd and thus laboratories may have had difficulties for sound calculation of the uncertainty budget of that measurand. For those laboratories which made the effort to provide an uncertainty estimate for the concentration of extractable Cd and additional assessment is given here which aims at giving the laboratory an indication of the plausibility of its uncertainty estimate. The standard uncertainty should fall in a range between a minimal required (u_{min}), and a maximal allowed (u_{max}) reported standard uncertainty. u_{min} is set to the standard uncertainty of a laboratory carrying the analysis on a routine basis is able to measure the measurand with a smaller uncertainty that the reference laboratory itself. u_{max} is set to the standard deviation accepted for the proficiency

test, $\hat{\sigma}$. If the standard uncertainty from the laboratory, $u_{lab} < u_{min}$ it is likely that the laboratory has underestimated its uncertainty. If $u_{lab} > u_{max}$, some effort should be made to reduce it because it exceeds the present state-of-the-art in that field of analysis.

The standard uncertainty of the laboratory (u_{lab}) was calculated dividing the reported expanded uncertainty by the reported coverage factor (*k*). When no uncertainty was reported, it was set to zero $(u_{lab} = 0)$. When *k* was not specified, the reported expanded uncertainty was considered as the half-width of a rectangular distribution; u_{lab} was then calculated by dividing this half-width by $\sqrt{3}$, as recommended by Eurachem and CITAC¹⁶.

8.3 Laboratory results and scorings

The results, as reported by the participants, are summarised in Table 2a-e for total Cd, total Pb, total Hg, extractable Cd and extractable Pb, respectively, together with the z- and zeta scores, where appropriate (not for total and extractable Pb). Laboratory codes were given randomly.

Three sets of figures are provided for total Cd, extractable Cd and total Hg (Fig 2-4). Each set includes (a) the Kernel Density plot, (b) individual mean value and associated expanded uncertainty, (c) the z- and zeta scores. For total and extractable Pb (Fig 5-6) only the mean values and associated expanded uncertainties are provided. The solid line represents the

assigned value, the dashed lines delimit the reference interval $(X_{ref} \pm 2u_{ref})$ and the dotted lines delimit the target interval $(X_{ref} \pm 2\sigma)$. The Kernel plots were obtained using a software tool developed by AMC¹⁷.

Taking into consideration the z-score, twenty-nine laboratories (94 %) performed well for total Cd against the target standard deviation of 16 %. Two laboratories (6 %) obtained z-scores 3 > |z|. For extractable Cd, twenty-two (84.5 %) out of the twenty-six laboratories which reported values for this measurand obtained z-scores $|z| \le 2$. Three laboratories (11.5 %) obtained a z-score 3 > |z| and one laboratory (4 %) obtained a z-score $2 < |z| \le 3$. For the determination of total Hg, twenty-four laboratories (92 %), obtained a satisfactory z-score against the target standard deviation of 22 % and two laboratories (8 %) obtained an unsatisfactory 3 > |z|, Table 3.

As regards zeta-scores for total Cd, twenty-five laboratories (81 %) obtained satisfactory zetascore, two laboratories (6 %) obtained a zeta-score $2 < |z| \le 3$ and four laboratories (13 %) obtained an unsatisfactory zeta-score. For total Hg, eleven laboratories (42 %) out of the twenty-six which reported data obtained a $|z| \le 2$, five laboratories (19 %) obtained a questionable zeta-score and 10 laboratories (39 %) obtained an unsatisfactory zeta-score |z| >3. One third of the laboratories seem to have problems in making an estimation of the uncertainty of their measurements which is consistent with the laboratory's deviation from the reference value for Hg, Table 3.

Additional information was gathered from the questionnaire completed by the participants. Thirteen laboratories have corrected their results for recovery and eighteen did not. Of those which did, nine used a reference material to calculate the recovery. The rest spiked the sample with a known amount of the same analyte to be measured. Of those which did not correct for recovery, four indicated that they do not do it in routine analysis, five replied that their recovery is 100 %, three declared not to have corrected for recovery because they do not have an adequate certified reference material to calculate it, two answered that corrections for recovery are not to be applied in analysis of heavy metals. Various reasons were given by the rest of the laboratories.

When asked about the level of confidence reflected by the reported coverage factor (k), twenty-three reported a level of 95 %, one of 90 %, one did not provide any figure, two gave an answer which did not correspond to the question, one indicated that no statistical analysis is applied, two said not to use coverage factor and one reported "*no idea*".

For uncertainty estimates, various combinations of two or more options (question 3 of the questionnaire shown in Annex 5) were given. Nine laboratories make uncertainty budget calculation according to ISO-GUM. Three use the known uncertainty of the standard method. Twenty-one report the uncertainty as estimated during in-house validation, ten use precision data and one makes a "guesstimate". Fifteen laboratories provide an uncertainty statement to their customers for this type of analysis.

Seventeen laboratories corrected their results for the water content. From those which did not, two indicated that it is not normally done according to their procedures. Four indicated that they did not correct for humidity because it was not requested by the organiser.

Two laboratories indicated to have introduced some modification to the prescribed protocol for the partial extraction of Cd and Pb: L24 took 1 g of test material, added 42.5 mL 5 %

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 HNO_3 and after extraction diluted to 50 mL. L17 took 1 g in 50 mL Erlenmeyer flask, and used paper filter "Red ribbon".

The thirty-one laboratories participating in this exercise apply total matrix digestion in routine analysis and no one of them applied partial extraction for routine analysis.

Seventeen laboratories analyse the test material following an official method. The information reported by the remaining fourteen laboratories about their method of analysis is summarised in Annex 6.

Twenty-three laboratories carry out this type of analysis on a routine basis. The distribution of them in terms of number of samples analysed per year in shown in Annex 7.

Twenty-nine laboratories have a quality system in place. One out the twenty-nine is accredited according to ISO 9000, the remaining twenty-eight being accredited according to ISO/IEC 17025. Twenty laboratories are accredited for this type of analysis (when applying total digestion of the matrix).

Twenty-five laboratories declared to participate regularly in ILC's for this type of analysis.

Twenty laboratories use a reference material for this type of analysis. Eight out of the twenty use the reference material for calibration purposes and seventeen use it for the validation of the method.

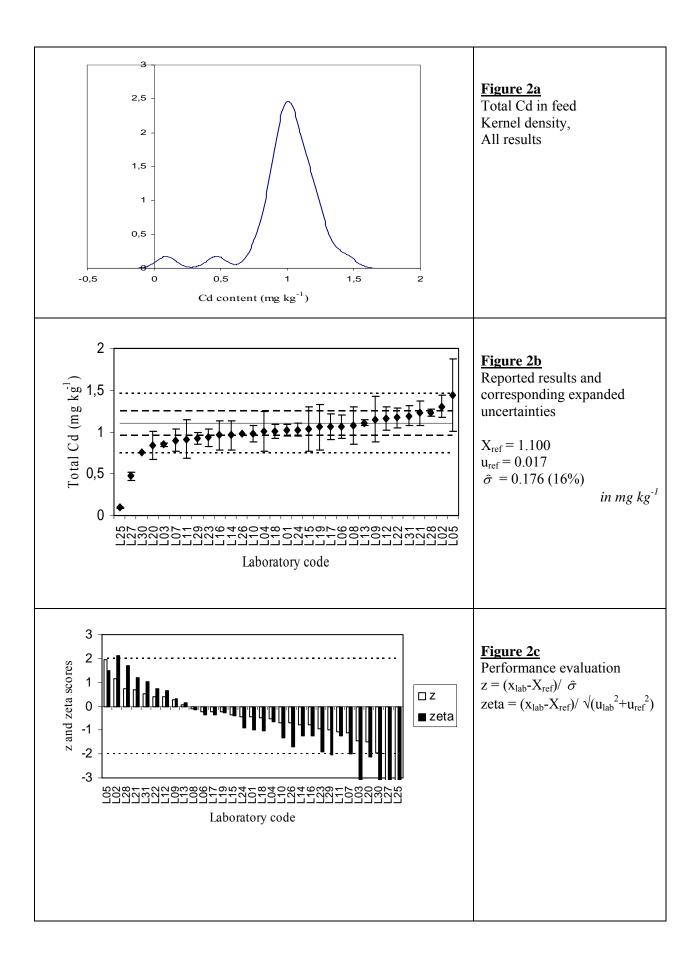
9 Conclusions

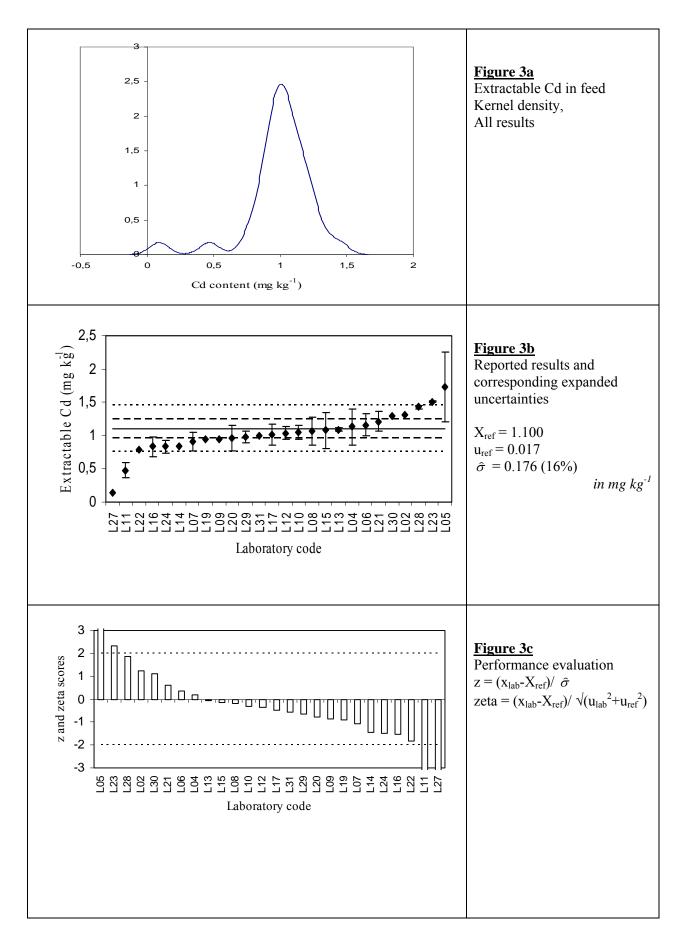
The main conclusion that can be extracted out of this study is that the concentration of total and extractable Cd (according to Directive 2002/32/EC) can be considered identical when analysing feed with a high content of organic matter. This implies that laboratories can continue applying total matrix digestion for the determination of Cd in organic feed matrices. Total matrix digestion helps to overcome some drawbacks of the partial extraction such as interferences from the matrix. BAM reported high carbon interferences when partial extraction was applied. Also L31 reported that extractable Pb could not be determined due to interferences.

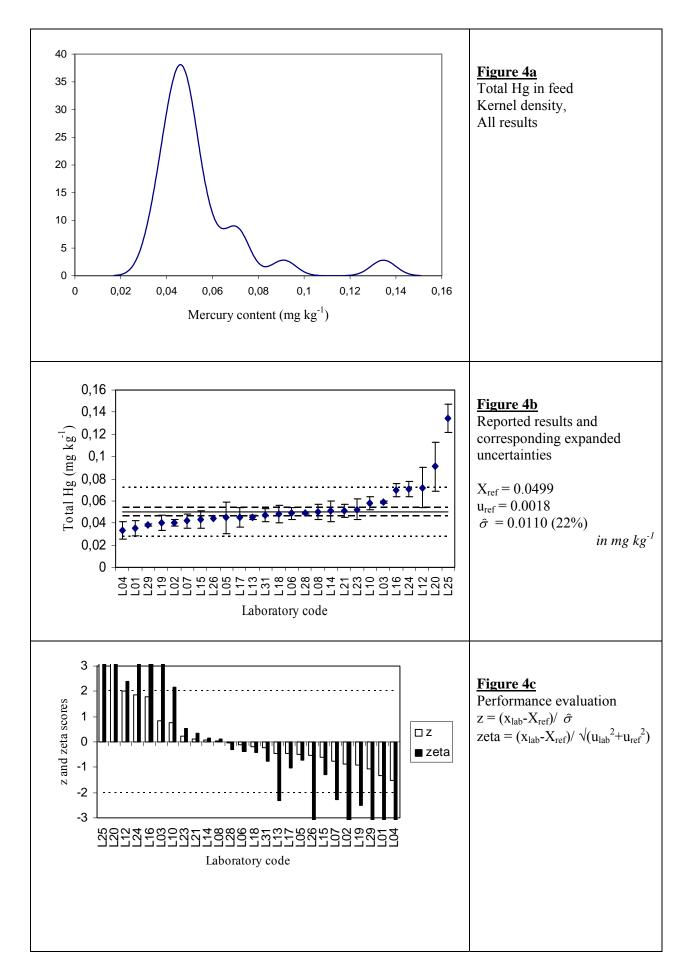
According to the Youden plot the concentration of total and extractable Pb are also similar. However, due to heterogeneity problems of the test material no definite conclusion can be extracted so far and further studies are needed.

Some attention must be paid to the fact that in general feed containing fish could suffer from a lack of homogeneity with the consequent impact in the results obtained. Nevertheless, further studies on this subject must be carried out before stating final conclusions. Nevertheless, it appears that the homogeneity of a test material may differ considerably for different measurands within the same batch. This shows that it may be insufficient to determine the homogeneity of only one measurand while assuming that the others behave similar.

Although the overall performance of the laboratories is quite satisfactory for total and extractable Cd and for total Hg, it must be pointed out that a number of laboratories have a problem with the estimation of their uncertainty in particular for the determination of total Hg.







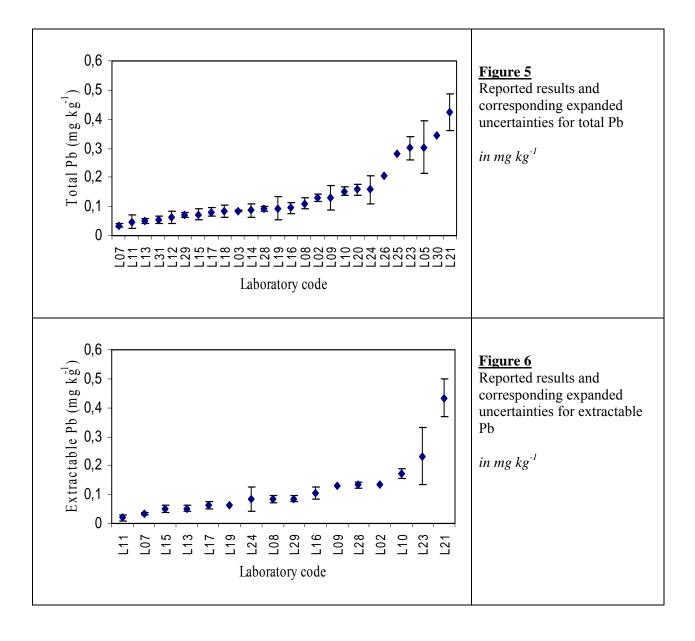


Table 2a: Total Cd, quantitative information reported by participants plus the laboratory scorings provided by the organiser.	
Total Cd content: $1.100 \pm 0.145 \text{ mg kg}^{-1}$	

Lab code	x1	x2	x3	x4	U	k	Technique	Mean-calc	Z	zeta
L01	1.01	1.02	1.04		0.07	3	ICP-AES	1.023	-0.4	-1.0
L02	1.41	1.18	1.33		0.131	2	ZETAAS	1.307	1.2	2.1
L03	0.8429	0.8459	0.8522		0.0156	2	AAS	0.847	-1.4	-3.5
L04	1.01	1	1.01		0.242	2	ICP-MS	1.007	-0.5	-0.7
L05	1.5	1.42	1.4		0.43	2	AAS	1.440	1.9	1.5
L06	1.07	1.056			0.14	2	ICP-AES	1.063	-0.2	-0.4
L07	0.904	0.894	0.904		0.135	2	ICP-AES	0.901	-1.1	-2.0
L08	1.07	1.08	1.09		0.22	2	ETAAS	1.080	-0.1	-0.2
L09	1.17	1.13			0.27	2	AAS	1.150	0.3	0.3
L10	1.05	0.95	0.93		0.098	b	AAS	0.977	-0.7	-1.3
L11	0.89	0.93	0.92		0.23	b	ETAAS	0.913	-1.1	-1.2
L12	1.18	1.25	1.07		0.14	2	AAS	1.167	0.4	0.7
L13	1.115	1.142	1.078		0.032	2	ICP-MS	1.112	0.1	0.2
L14	0.948	0.946	0.991		0.173	2	ETAAS	0.962	-0.8	-1.2
L15	1.03	1.035	1.043		0.26	b	ETAAS	1.036	-0.4	-0.4
L16	0.93	0.992			0.173	2		0.961	-0.8	-1.2
L17	1.064	1.06			0.15	2	ICP-MS	1.062	-0.2	-0.4
L18	1.018	0.955	1.011	1.064	0.085	2	ICP-MS	1.012	-0.5	-1.0
L19	1.02	1.1			0.27	2	ETAAS	1.060	-0.2	-0.3
L20	0.929	0.759	0.823		0.172	b		0.837	-1.5	-2.1
L21	1.23	1.22			0.15	2	ICP-QMS	1.225	0.7	1.2
L22	1.131	1.205			0.12	2	AAS	1.168	0.4	0.7
L23	0.88	0.981			0.1	2	AAS	0.931	-1.0	-1.9
L24	1.051	1.04	0.981		0.078	b	AAS	1.024	-0.4	-0.9
L25	0.08	0.08	0.11	0.1	0.01	b	AAS	0.093	-5.7	-13.8
L26	0.976	0.965	0.987	0.972	а	2	ICP-MS	0.975	-0.7	-1.7
L27	0.466	0.473	0.471		0.054	2	GF-AAS	0.470	-3.6	-8.1
L28	1.16	1.27	1.26		0.037	b	ICP-AES	1.230	0.7	1.7

Lab code	x1	x2	x3	x4	${f U}$	k	Technique	Mean-calc	Z	zeta
L29	0.927	0.929	0.934		0.07	b	ICP-MS	0.930	-1.0	-2.0
L30	0.7892	0.7668	0.7204		a	b	ICP-OES	0.759	-1.9	-4.7
L31	1.18	1.17	1.22	1.21	0.12	2	AAS	1.195	0.5	1.0

All results expressed in (mg kg⁻¹) a) k not reported; $u = U/\sqrt{3}$

b) u not reported; set to zero in zeta.

Lab code	x1	x2	x3	x4	U	k	Technique	Mean-calc	Z
L02	1.31	1.32			a	b	ZETAAS	1.315	1.2
L04	1.13	1.13	1.13		0.271	2	ICP-MS	1.130	0.2
L05	1.68	1.8	1.72		0.52	2	AAS	1.733	3.6
L06	1.15	1.17			0.16	2	ICP-AES	1.160	0.3
L07	0.916	0.909	0.914		0.137	2	ICP-AES	0.913	-1.1
L08	1.04	1.07	1.1		0.21	2	AAS	1.070	-0.2
L09	0.963	0.932	0.958		a	b	AAS	0.951	-0.8
L10	1.01	1.05	1.15	0.98	0.098	b	AAS	1.048	-0.3
L11	0.47	0.49	0.48		0.12	b	ETAAS	0.480	-3.5
L12	1.06	1.04	1.02		0.09	2	AAS	1.040	-0.3
L13	1.078	1.115	1.076		0.022	2	ICP-MS	1.090	-0.1
L14	0.846	0.856	0.84		a	b		0.847	-1.4
L15	1.063	1.058	1.105		0.27	b	ETAAS	1.075	-0.1
L16	0.83	0.831			0.15	2		0.831	-1.5
L17	1.022	1.005			0.15	2	ICP-MS	1.014	-0.5
L19	1.02	0.86			a	b	ETAAS	0.940	-0.9
L20	0.893	0.924	1.077		0.197	b		0.965	-0.8
L21	1.25	1.17			0.15	2	ICP-QMS	1.210	0.6
L22	0.796	0.765			a	b	AAS	0.781	-1.8
L23	1.511	1.509			0.01	2	AAS	1.510	2.3
L24	0.839	0.895	0.775		0.098	b	AAS	0.836	-1.5
L27	0.148	0.128	0.121		a	b	GF-AAS	0.132	-5.5
L28	1.36	1.42	1.51		0.026	b	ICP-AES	1.430	1.9
L29	0.978	0.992	0.977	0.984	0.09	b	ICP-MS	0.983	-0.7
L30	1.2575	1.34			a	b	ICP-OES	1.299	1.1
L31	1.01	1.02	0.98	0.98	a	b	AAS	0.998	-0.6

Table 2b: Extractable Cd, quantitative information reported by participants plus the laboratory scorings provided by the organiser.**Extractable Cd content:** $1.100 \pm 0.145 \text{ mg kg}^{-1}$

All results expressed in (mg kg⁻¹). a) k not reported; $u = U/\sqrt{3}$, b) u not reported.

	$\frac{10 \text{tai Hg content: } 0.0499 \pm 0.0057 \text{ Hg kg}}{10 \text{tai Hg content: } 0.0499 \pm 0.0057 \text{ Hg kg}}$									
Lab code	x1	x2	x3	x4	\mathbf{U}	k	Technique	Mean-calc	Z	zeta
L01	0.032	0.037	0.037		0.007	3	CV-ICP-AES	0.035	-1.3	-4.9
L02	0.04	0.041	0.04		0.003	2	TDA-AAS	0.040	-0.9	-4.0
L03	0.06	0.0587	0.059	0.0593	0.0009	2	HG-AAS	0.059	0.9	4.9
L04	0.06	0.02	0.02		0.0075	2	ICP-MS	0.033	-1.5	-4.0
L05	0.047	0.042	0.045		0.014	2	AMA 254	0.045	-0.5	-0.7
L06	0.0493	0.0481			0.0051	2	CV-AAS	0.049	-0.1	-0.4
L07	0.0418	0.0399	0.0435		0.0062	2	CV-AAS	0.042	-0.7	-2.3
L08	0.05	0.05	0.051		0.007	2	CV-AAS	0.050	0.0	0.1
L10	0.059	0.061	0.055	0.058	0.0058	b	CVAAS	0.058	0.8	2.2
L11	< 0.1	< 0.1	< 0.1				HG-AAS			
L12	0.062	0.066	0.088		0.018	2	AAS	0.072	2.0	2.4
L13	0.047	0.043	0.045		0.002	2	AMA	0.045	-0.4	-2.3
L14	0.051	0.049	0.052		0.009	2	AAS	0.051	0.1	0.2
L15	0.047	0.046	0.037		0.008	b	CV-AAS	0.043	-0.6	-1.3
L16	0.066	0.073			0.006	2		0.070	1.8	5.6
L17	0.045	0.045			0.009	2		0.045	-0.4	-1.0
L18	0.045	0.05	0.045	0.052	0.008	2	ICP-MS	0.048	-0.2	-0.4
L19	0.04	0.04			0.007	2	AAS	0.040	-0.9	-2.5
L20	0.119	0.063			0.022	b		0.091	3.7	3.1
L21	0.0501	0.0522			0.0061	2	CV-AAS	0.051	0.1	0.4
L22	<0.1						ICP-MS			
L23	0.0569	0.0479			0.009	2	HG-AAS	0.052	0.2	0.5
L24	0.067	0.07	0.074		0.007	b	AAS	0.070	1.9	4.5
L25	0.139	0.13			0.013	b	AMA 254	0.135	7.7	10.8
L26	0.044	0.0437	0.0444	0.0438	a	2	AMA-254	0.044	-0.5	-3.2
L28	0.05	0.05	0.048		0.001	b	AFS	0.049	-0.1	-0.3
L29	0.0373	0.037	0.0395	0.0398	0.001	b	Combustion in oxygen + CV - AAS	0.038	-1.0	-5.9
L31	0.048	0.047	0.048	0.046	0.006	2	AAS	0.047	-0.2	-0.8

Table 2c: Total Hg, quantitative information reported by participants plus the laboratory scorings provided by the organiser. **Total Hg content:** $0.0499 \pm 0.0037 \text{ mg kg}^{-1}$

All results expressed in (mg kg⁻¹). a) k not reported; $u = U/\sqrt{3}$, b) u not reported; set to zero in zeta.

Lab code	x1	x2	x3	x4	U	k	Technique	Mean-calc
L01	< 0.78	< 0.78	< 0.78				ICP-AES	
L02	0.11	0.15	0.13		0.013	2	ZETAAS	0.130
L03	0.0854	0.0876	0.0835		0.0021	2	AAS	0.086
L04	< 0.1	< 0.1	< 0.1				ICP-MS	
L05	0.31	0.25	0.35		0.09	2	AAS	0.303
L06	< 0.5	< 0.5	< 0.5				ICP-AES	
L07	0.0324	0.0357	0.0388		0.0053	2	ETAAS	0.036
L08	0.107	0.117	0.108		0.018	2	ETAAS	0.111
L09	0.13	0.13			0.04	2	AAS	0.130
L10	0.157	0.16	0.15	0.145	0.0145	b	AAS	0.153
L11	0.05	0.04	0.05		0.023	b	ETAAS	0.047
L12	0.069	0.062	0.057		0.021	2	AAS	0.063
L13	0.058	0.044	0.047		0.007	2	ICP-MS	0.050
L14	0.086	0.084	0.09		0.022	2	ETAAS	0.087
L15	0.086	0.065	0.066		0.018	b	ETAAS	0.072
L16	0.084	0.107			0.019	2		0.096
L17	0.083	0.079			0.015	2	ICP-MS	0.081
L18	0.077	0.081	0.089	0.09	0.02	2	ICP-MS	0.084
L19	0.1	0.1	0.08		0.04	2	ETAAS	0.093
L20	0.168	0.154	0.151		0.018	b	AAS	0.158
L21	0.411	0.433			0.063	2	ICP-QMS	0.422
L22	< 0.3						AAS	
L23	0.271	0.331			0.04	2	AAS	0.301
L24	0.183	0.135	0.157		0.048	b	AAS	0.158
L25	0.28	0.21	0.35		a	b	AAS	0.280
L26	0.209	0.203	0.204	0.21	a	2	ICP-MS 7500c	0.207
L27	< 0.023	< 0.023	< 0.023				GF-AAS	
L28	0.087	0.093	0.094		0.009	b		0.091
L29	0.0725	0.068	0.0706	0.0729	0.007	b	ICP-MS	0.071

Table 2d: Total Pb, quantitative information reported by participants.

Lab code	x1	x2	x3	x4	\mathbf{U}	k	Technique	Mean-calc
L30	0.3822	0.3066	0.3374		a	b	ICP-OES	0.342
L31	0.052	0.057	0.051		0.012	2	AAS	0.053

All results expressed in (mg kg⁻¹). a) k not reported, b) u not reported.

Lab code	x1	x2	x3	x4	\mathbf{U}	k	Technique	Mean-calc
L02	0.12	0.15			a	b	ZETAAS	0.135
L04	< 0.1	< 0.1	< 0.1				ICP-MS	
L05	< 0.2	< 0.2	< 0.2				AAS	
L06	< 0.5	< 0.5					ICP-AES	
L07	0.0283	0.0349	0.0365		0.005	2	ETAAS	0.033
L08	0.073	0.1	0.077		0.014	2	ETAAS	0.083
L09	0.13	0.13			a	b	AAS	0.130
L10	0.17	0.18	0.16	0.175	0.0175	b	Z-ETA-AAS	0.171
L11	0.02	0.02	0.02		0.01	b	ETAAS	0.020
L13	0.043	0.062	0.051		0.01	2	ICP-MS	0.052
L14	< 0.005	< 0.005	< 0.005				ETAAS	
L15	0.073	0.042	0.039		0.012	b	ETAAS	0.051
L16	0.105	0.107			0.021	2		0.106
L17	0.073	0.05			0.012	2	ICP-MS	0.062
L19	0.062	0.064			a	b	ETAAS	0.063
L20	< 0.05	< 0.05	< 0.05					
L21	0.447	0.42			0.065	b	ICP-QMS	0.434
L22	< 0.3						AAS	
L23	0.285	0.18			0.1	2	AAS	0.233
L24	0.081	0.065	0.103		0.043	b	AAS	0.083
L27	< 0.023	< 0.023	< 0.023				GF-AAS	
L28	0.12	0.14	0.14		0.011	b	AAS	0.133
L29	0.0859	0.0894	0.083	0.085	0.01	b	ICP-MS	0.086

Table 2e: Extractable Pb, quantitative information reported by participants.

All results expressed in (mg kg⁻¹). a) k not reported, b) u not reported.

	Total Cd (%)	Extractable Cd (%)	Total Hg(%)
z-score			
Satisfactory	94	84.5	92
Questionable	0	4	0
Unsatisfactory	6	11.5	8
zeta-score			
Satisfactory	81		42
Questionable	6		19
Unsatisfactory	13		39

Table 3: Percentages of laboratories scoring satisfactory, questionable and unsatisfactory.

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Organisation	Country
AGES Zemtrum Analytik und Mikrobiologie	Austria
AGES Competence Centre of Elements	Austria
• Institute of Public Health	Belgium
National Veterinary Services	Bulgaria
• The State Veterinary Institute (SVI) in Olomuc	Czech Republic
State General Laboratory	Cyprus
• The Danish Plant Directorate	Denmark
• The Veterinary and Food Laboratory	Estonia
Agricultural Research Centre	Estonia
Finnish Food Safety Authority. Evira	Finland
SCL. Laboratoire de Bordeaux Pessac	France
Ministry of Rural Development and Food	Greece
General Chemical State Laboratory	Greece
Regional Centre of Magnisia	Greece
• Bundesamt für Verbraucherschutz und Lebensmittelsicherheit (BVL)	Germany
National Institute for Agricultural Quality Control	Hungary
Public Analyst's Laboratory	Ireland
Istituto Zooprofilattico Turin	Italy
National Veterinary Laboratory	Lithuania

Food Control Laboratory	Latvia
Public Health Laboratory	Malta
Instituut voor voedselveiligheid (RIKILT)	The Netherlands
Voedsel en Waren Autoriteit (VWA)	The Netherlands
National Veterinary Research Institute	Poland
IPIMAR – Instituto de Investigação das Pescas e do Mar	Portugal
Laboratório Nacional de Investigação Veterinária	Portugal
HVPHI	Romania
State Veterinary and Food Institute	Slovakia
National Veterinary Institute	Slovenia
Laboratorio Arbitral Agroalimentario	Spain
National Veterinary Institute	Sweden
	Public Health Laboratory Instituut voor voedselveiligheid (RIKILT) Voedsel en Waren Autoriteit (VWA) National Veterinary Research Institute IPIMAR – Instituto de Investigação das Pescas e do Mar Laboratório Nacional de Investigação Veterinária HVPHI State Veterinary and Food Institute National Veterinary Institute Laboratorio Arbitral Agroalimentario

Countries not appearing of the above list did not register to this interlaboratory comparison.

11 References

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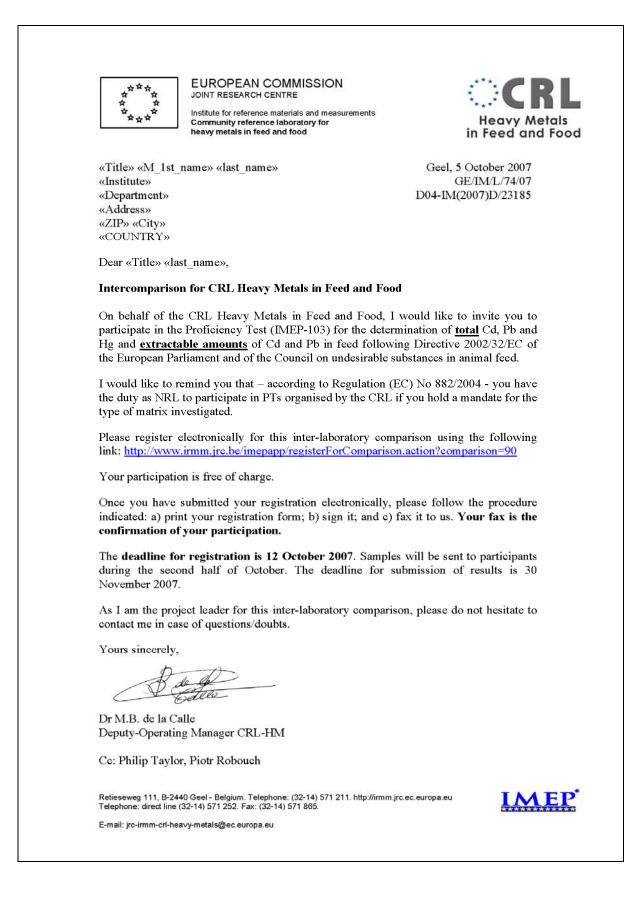
¹⁶ Eurachem/CITAC guide "Quantifying Uncertainty in Analytical Measurements" (2000), see www.eurachem.ul.pt

¹⁷ The software to calculate Kernel densities is provided by the Statistical Subcommittee of the Analytical Methods Committee (AMC) of the Royal Society of Chemistry and described in the AMC Technical Brief "Representing data distributions with Kernel density estimates" (2006), see <u>www.rsc.org/amc</u>.

Annexes

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Annex 1: Invitation letter to laboratories



Annex 2: Results of the homogeneity study

1a. Homogeneity data for total Cd in feed.

According to the IUPAC International Harmonised Protocol¹¹

	Cd (µg kg ⁻¹)		
Bottle ID	Replicate 1	Replicate 2	
15	880,16	889,94	
32	869,04	876,81	
52	827,85	834,54	
80	824,08	815,92	
96	811,63	790,05	
112	790,85	773,39	
138	784,69	758,19	
142	767,36	769,05	
175	754,31	754,91	
195	733,05	734,5	
Mean, n	802,016	20	
Target RSD %	16		
S_{an}^{2}	87,2	7398	
S_{sam}^2	2453,812114		
σ_{all}^2	38,9779776		
Critical	2944,398267		
S _{sam} ² <critical?< td=""><td colspan="3">ACCEPT</td></critical?<>	ACCEPT		

According to ISO 13528³

0.3σ	38,9779776
S _X	49,97448453
S _w	9,342054378
S _s	49,53596789
s _s ≤ 0.3 σ	NOT ACCEPT

1b. Stability data for total Cd in feed.

As computed by SOFT CRM

		Total C	'd at 4•C		
			Time in	Weeks	
samples		0	3	5	8
	1	914,2	824,1	812,5	787,8
	2	836	815,5	804,2	801,9
Slope =		-9,778			
SE Slope	e =	3,198			
Intercept	; =	863,637			
SE Inter	rent =	15 828	2		

SE Intercept = 15,828 Correlation Coefficient =0,609

Slope of the linear regression significantly > 0 (95%):Yes

Slope of the linear regression significantly > 0 (99%)

:No

Calculation of u _{sts}	
Given Xshelf = 6 Weeks	
$u_{sts} = 19,186$	
u_{sts} [%] = 2,3%	

2. Homogeneity data for total Pb in feed.

According to the IUPAC International Harmonised Protocol¹¹

	Pb (µg kg ⁻¹)		
Bottle ID	Replicate 1	Replicate 2	
15	75,04	78,05	
32	69,3	70,61	
52	113,09	76,89	
80	76,94	99,85	
96	149,79	96,96	
112	106,79	81,42	
138	131,78	81,54	
142	78,95	83,41	
175	83,9	88,81	
195	81,58	83,98	
Mean, n	90,434	20	
Target RSD %	22		
S_{an}^2	392,72737		
S_{sam}^2	43,62916944		
σ_{all}^2	35,6247112		
Critical	463,6291008		
S _{sam} ² <critical?< td=""><td colspan="3">ACCEPT</td></critical?<>	ACCEPT		

According to ISO 13528³

0.3σ	5,968644
S _x	15,49170276
S _w	19,81735023
S _S	6,605238031
s _s ≤ 0.3 σ	NOT ACCEPT

2b. Stability data for total Pb in feed.

As computed by SOFT CRM

			Time in	Weeks	
samples		0	3	5	8
	1	111,3	68,5	74	128,7
	2	79,2	68	93,2	56,8

Slope =	0,157	
SE Slope =	3,200	
Intercept =	84,333	
SE Intercept =	15,839	
Correlation Coe	fficient =0,00	00
Slope of the line	ar regression	significantly > 0 (95%)
:No		
Slope of the line	ar regression	significantly > 0 (99%)
:No		

Calculation of u _{sts}
Given Xshelf = 6 Weeks
$u_{sts} = 17,780$
$u_{sts} [\%] = 20,9\%$

3. Homogeneity data for total Hg in feed.

According to the IUPAC International Harmonised Protocol¹¹

	Hg (µg kg ⁻¹)		
Bottle ID	Replicate 1	Replicate 2	
112	45,68	45,2	
138	44,35	45,75	
175	45,4	45,72	
32	45,68	45,15	
80	45,13	45,66	
142	45,73	45,3	
96	45,54	45,44	
195	44,81	46,17	
52	46,14	45,76	
15	45,07	44,67	
Mean, n	45,4175	20	
Target RSD %	2	2	
S_{an}^2	0,260175		
S_{sam}^2	-0,045675		
σ_{all}^2	8,985335978		
Critical	17,15520839		
S _{sam} ² <critical?< td=""><td colspan="3">ACCEPT</td></critical?<>	ACCEPT		

According to ISO 13528³

0.3σ	2,997555
S _x	0,290538294
S _w	0,510073524
S _S	
s _s ≤0.3 σ	ACCEPT

3b. Stability data for total Hg in mineral water.

As computed by SOFT CRM

Total Hg at 4°C

	Time in Weeks				Time in Weeks			Calculation of u _{sts}
samples	0	3	5	8	Given Xshelf = 6 Weeks			
1	45,51	45,36	45,93	45,5	$u_{sts} = 0,287$			
2	45,27	45,33	44,77	46,03	$u_{sts}[\%] = 0.6\%$			
Slope = SE Slope = Intercept = SE Intercept = Correlation Coe Slope of the line :No Slope of the line :No	0,240 fficient =0, ar regressio	on signific	2	× /				

4a. Homogeneity data for extractable Cd in feed.

According to the IUPAC International Harmonised Protocol¹¹

	Cd (µg kg ⁻¹)			
Bottle ID	Replicate 1	Replicate 2		
15	924,4	1024,73		
32	1004,46	1037,49		
52	975,56	968,12		
80	957,06	944,77		
96	932,97	912,99		
112	908,83	884,52		
138	877,01	878		
142	866,88	862,1		
175	866,56	869,25		
195	857,74	856,71		
Mean, n	920,5075	20		
Target RSD %	1	6		
S _{an} ²	619,289475			
S_{sam}^2	2902,416142			
σ_{all}^2	2001,369151			
Critical	4388,056373			
S _{sam} ² <critical?< td=""><td>ACC</td><td>CEPT</td></critical?<>	ACC	CEPT		

According to ISO 13528³

11000141118 10 180 10020	
0.3σ	44,7366645
S _X	56,67504635
Sw	24,88552742
S _S	53,87407671
$s_s \le 0.3 \sigma$	NOT ACCEPT

4b. Stability data for extractable Cd in feed.

As computed by SOFT CRM

Extractable Cd at 4°C

Time in Weeks					Calculation of u _{sts}	
samples		0	3	5	8	Given Xshelf = 6 Weeks
	1	835,27	850,48	822,76	819,86	$u_{sts} = 11,313$
~	2	839,78	804,15	823,26	809,11	$u_{sts} [\%] = 1,4\%$

Slope =	-2,837	
SE Slope =	1,675	
Intercept =	836,933	
SE Intercept =	8,291	
Correlation Coef	ficient =0,32	3
Slope of the linea	ar regression	significantly > 0 (95%)
:No		
Slope of the linea	ar regression	significantly $<> 0$ (99%)
:No		

5. Homogeneity data for extractable Pb in feed.

According to the IUPAC International Harmonised Protocol¹¹

	Рb (µg kg ⁻¹)			
Bottle ID	Replicate 1	Replicate 2		
15	61,05	113,7		
32	68,53	71,21		
52	62,98	74,75		
80	71,16	69,52		
96	109,3	103,08		
112	133,28	78,5		
138	87,31	149,78		
142	94,68	79,63		
175	73,22	71,86		
195	77,89	64,95		
Mean, n	85,819	20		
Target RSD %	2	2		
S_{an}^2	512,9	01304		
S_{sam}^2	83,33136778			
σ_{all}^2	32,08150771			
Critical	578,3554049			
S _{sam} ² <critical?< td=""><td>ACC</td><td>CEPT</td></critical?<>	ACC	CEPT		

According to ISO 13528³

0.3σ	5,664054
S _x	18,43333632
S _w	22,64758354
S _S	9,128601633
s _s ≤ 0.3 σ	NOT ACCEPT

5b. Stability data for extractable Pb in feed.

As computed by SOFT CRM

Extractable Pb at 4°C

Time in Weeks						
samples	0	3	5	8		
1	100,59	158,58	94,58	100,68		
2	74,86	75,24	174,69	114,67		
Slope =	2,868					
Slope = SE Slope =	2,868 4,65					
Intercept =	100,2					
SE Intercept	,					
Correlation C	Coefficient =	=0,060				
Slope of the	linear regres	ssion signif	ficantly <> () (95%)		
:No						
Slope of the 1	linear regres	ssion signif	ficantly <> 0) (99%)		
:No						

Calculation of u _{sts}	
Given Xshelf = 6 Weeks	
$u_{sts} = 26,661$	
u_{sts} [%] = 23,9%	

Annex 3: Letter accompanying the sample

	CRI
	**** Heavy Metals in Feed and Foo
Geel, 16 Oc GE/IM/L/7	ctober 2007
«ADDRESS2» «ADDRESS3» «ADDRESS4» «ZIP» «TOWN» «COUNTRY»	
Participation to IMEP-103, a proficiency test exercise for total Cd, Pb and Hg and <u>extractable amounts</u> of Cd and Pb in	
Dear «TITLE» «SURNAME»,	
Thank you for participating in the IMEP-103 intercomparison total Cd, Pb and Hg and extractable amounts of Cd and Pb in place in the frame of the CRL Heavy Metals in Feed and Food.	
<u>This parcel contains</u> : a) One glass bottle containing approximately 20 g of the test mat b) A "Confirmation of Receipt" form c) This accompanying letter	terial
Please check whether the bottle containing the test material rem transport. Then fax (at +32-14-571865) or send the "Confirmation You should store the samples in a dark and cold place (4 $^{\circ}$ C) until	on of receipt" form back.
The measurands are: total Cd, Pb and Hg and extractable according to Directive 2002/32/EC of the European Parliamen undesirable substances in animal feed, in a feed matrix of fish or	it and of the Council on
As agreed upon during the workshop held in September, th <u>extractable</u> amounts of Cd and Pb shall be carried out by strictl procedure:	
Retieseweg 111, B-2440 Geel - Belgium. Telephone: (32-14) 571 211. http://imm.jrc.ec Telephone: direct line (32-14) 571 252. Fax: (32-14) 571 865.	«PARTKEY» .europa.eu

Protocol for the partial extraction of Cd and Pb in feed (IMEP-103) 1. Weigh about 2 g of the prepared test sample to the nearest 1 mg into a 250 mL beaker. 2. Add 85 mL of a 5 % (w/w) HNO_3 solution (see note for the preparation of the HNO₃ solution). 3. Cover the beaker with a watch-glass and boil for 30 min on a hot plate (make sure that the plate warms up homogeneously all over the surface). 4. Allow to cool. Decant the liquid into a 100 mL volumetric flask, rinsing the beaker and the watch-glass several times with 5 % (w/w) HNO₃. 5. Dilute to the mark with 5 % (w/w) HNO₃. 6. After homogenising, filter through a fry folded filter paper into a dry container. Use the first portion of the filtrate to rinse the glassware and discard that part. If the determination is not carried out immediately, the container with filtrate shall be stoppered. 7. Carry out a blank test at the same time as the extraction, with only the reagents and follow the same procedure as for the samples. To construct the calibration curve dilute the standards in 5 % (w/w) HNO₃. NOTE: To prepare 1 kg stock of 5 % (w/w) HNO_3 (density ~ 1.0257 kg/l): mix 77 g of 65 % (w/w) HNO₃ with 923 g water. Use a balance of two digits for the weighing. For the determination of the total content of Cd, Pb and Hg the procedure that you use should resemble as closely as possible the one that you use in routine sample analysis. Please perform two or three independent measurements per parameter. Correct the measurement results for recovery, and report the corrected values, plus their mean on the reporting website. The results should be reported in the same form (e.g., number of significant figures) as those normally reported to the customer. You can find the reporting website at www.irmm.jrc.be/imepapp/jsp/loginResult.jsp To access this webpage you need a personal password key, which is: «PARTKEY». The system will guide you through the reporting procedure. Please enter for each parameter the two or three measurement results plus the technique you used, but do not report the uncertainty for each individual measurement. In addition, please report the mean of the results with technique and with uncertainty information in the allocated space for "measurement 4". After entering all results, please also complete the relating questionnaire. Do not forget to submit and confirm always when required. Directly after submitting your results and the questionnaire information online, you will be prompted to print the completed report form. Please do so, sign the paper version and return it to IRMM by fax (at +32-14-571-865) or by e-mail. Check your results carefully for any errors before submission, since this is your definitive confirmation. The deadline for submission of results is 30/11/2007. 2 «PARTKEY»

Please keep in mind that collusion is contrary to professional scientific conduct and serves only to nullify the benefits of proficiency tests to customers, accreditation bodies and analysts alike.

Your participation in this project is greatly appreciated. If you have any remaining questions, please contact me by e-mail: <u>JRC-IRMM-CRL-HEAVY-METALS@ec.europa.eu</u>

With kind regards

Dr. M.B. de la Calle IMEP-103 Co-ordinator

Enclosures: 1) one glass bottle containing the test material; 2) confirmation of receipt form; 3) Accompanying letter.

Cc: P. Taylor

«PARTKEY»

3

Annex 4: Sample receipt confirmation form

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Date of package arrival	Date of package arrival	ANY REMARKS		
Signature Please return this form to: Dr. Beatriz de la Calle IMEP-103 Coordinator EC-JRC-IRMM Retieseweg 111 B-2440 GEEL, Belgium Fax : +32-14-571865 e-mail : JRC-IRMM-CRL-HEAVY-METALS@ec.europa.eu Retieseweg 111, B-2440 Geel - Belgium. Telephone: (32-14) 571 211. http://mmm.jrc.ec.europa.eu Retieseweg 111, B-2440 Geel - Belgium. Telephone: (32-14) 571 211. http://mmm.jrc.ec.europa.eu	Signature Please return this form to: Dr. Beatriz de la Calle IMEP-103 Coordinator EC-JRC-IRMM Retieseweg 111 B-2440 GEEL, Belgium Fax : +32-14-571865 e-mail : JRC-IRMM-CRL-HEAVY-METALS@ec.europa.eu Retieseweg 111, B-2440 Geel - Belgium. Telephone: (32-14) 571 211. http://immm.jrc.ec.europa.eu Retieseweg 111, B-2440 Geel - Belgium. Telephone: (32-14) 571 211. http://immm.jrc.ec.europa.eu			
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E-mail: jrc-imm-crl-heavy-metals@ec.europa.eu	E-mail: jrc-imm-crl-heavy-metals@ec.europa.eu	Telephone: direct line (32-14) 571 2	jium. Telephone: (32-14) 571 211. http://imm.jrc.ec.e 52. Fax: (32-14) 571 865.	
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Annex 5: Questionnaire

	European Com Joint Research Institute for Re	Centre	faterials and Measurements	IRMM Interla	boratory Comparison	
	> Results >	Questic	onnaire			
	ctions			Questionnaire for IME	EP-103	
sults	5					
		1.	Did you apply a recovery factor to Ves No	o correct your measurement results?		
			If Yes, what are the recovery facto	ors (R, in %) you used:		
			for Cd (in %)			
			for Hg (in %)			
			for Pb (in %)			
			If Yes, did you determine R by:			
			using a reference material		Ε	
			adding a known amount of the s	ame parameter(s) to be measured ("spikin	ng")	
			other		E	
			If other, please specify			
			If No, please state why:			
			ii ivo, piease state wity.			
		2.	What is the level of confidence r	eflected by the coverage (k) factors state	ed above? (in %)	
		3.	What is the basis of your uncerta	ainty estimate (multiple answers are pos	sible)	
			uncertainty budget according to I	SO-GUM		
			known uncertainty of the standar	d method		
				ermined during in-house validation		
			measurement of replicates (i.e.)	precision)		
			expert guesstimate use of intercomparison data			
			other			
			If other, please specify.			
		4.	Do you usually provide an uncert	ainty statement to your custumers for thi	is type of analysis?	
			🔿 Yes 🔿 No			
		5.	Did you correct for the water co	ntent of the sample?		
				Yes	No	
				0	0	
			If Yes, what is the water conten	t (in % of the sample mass)?		
			In the second se			
			If No, what was the reason not	to do this?		
			Did you modify the properihed or	rational for the vertical diversion?		
		6.	Did you modify the prescribed pr Ves O No	ococorror the partial digestion?		
			If yes, please specify the modific	ations introduced.		
		_				
		7.	Did you analyse the sample acco Ves O No	ording to an official method?		
				50 characters for each reply) your:		
			sample pre-treatment			
			digestion step			

CRL-HM in Feed and Food. Total Cd, Pb and Hg and extractable Cd and Pb in feed

	instrument calibration step						
	If Yes, which:						_
8.	Does your laboratory carry ou	ut this type of analysis (a	is regards the pai	rameters, matrix a	nd methods) on	a routir	ne basis?
			0-50 samples per year	50-250 samples peryear	250-1000 sam peryear		more than 10 samples per ye
	If Yes, please estimate the nu Hg, Pb measurements togeth		0	0	0		0
9.	Does your laboratory have a c	quality system in place?					
		ISO 9000 series		ISO/IEC 17025	0	Other	
	If Yes, wich:						
	If Other, please specify.						
10.	Which type of sample treatm	ent do you routinely use	for such samples	\$?			
	Total digestion						
	Partial digestion (according to						
11.	Is your laboratory accredited	for the sample treatmer	nt that you specify	/ in question 10?			
11.	Is your laboratory accredited	for the sample treatme	nt that you specify	/ in question 10?			
11. 12.	Does your laboratory take pa				m a regular bas	sis?	
	Does your laboratory take part				n a regular bas	sis?	
	Does your laboratory take pa				m a regular bas	sis?	
	Does your laboratory take par Yes No If yes, which one(s): Does your laboratory use a re	rt in an interlaboratory c	omparison for thi	is type of analysis c	m a regular bas	sis?	
12.	Does your laboratory take par Yes No If yes, which one(s):	rt in an interlaboratory c	omparison for thi	is type of analysis c	n a regular bas	sis?	
12.	Does your laboratory take par Yes No If yes, which one(s): Does your laboratory use a re	rt in an interlaboratory c 	omparison for thi s type of analysis	is type of analysis c			
12.	Does your laboratory take par Yes No If yes, which one(s): Does your laboratory use a re Yes No	rt in an interlaboratory c eference material for this	omparison for thi s type of analysis ures?	is type of analysis c	Yes	No	
12.	Does your laboratory take par Yes No If yes, which one(s): Does your laboratory use a re Yes No If YES, is the material used for	rt in an interlaboratory c eference material for this	omparison for thi s type of analysis ures?	is type of analysis c	Yes	No O	
12.	Does your laboratory take par Yes No If yes, which one(s): Does your laboratory use a re Yes No If YES, is the material used fo If YES, is the material used fo If YES, which one(s):	rt in an interlaboratory c eference material for this or the validation of proced or calibration of instrumer	omparison for thi s type of analysis ures?	is type of analysis c	Yes	No O	
12.	Does your laboratory take par Yes No If yes, which one(s): Does your laboratory use a re Yes No If YES, is the material used fo If YES, is the material used fo	rt in an interlaboratory c eference material for this or the validation of proced or calibration of instrumer	omparison for thi s type of analysis ures?	is type of analysis c	Yes	No O	
12.	Does your laboratory take par Yes No If yes, which one(s): Does your laboratory use a re Yes No If YES, is the material used fo If YES, is the material used fo If YES, which one(s):	rt in an interlaboratory c eference material for this or the validation of proced or calibration of instrumer	omparison for thi s type of analysis ures? nts?	is type of analysis c	Yes	No O	
12. 13.	Does your laboratory take par Yes No If yes, which one(s): Does your laboratory use a re Yes No If YES, is the material used fo If YES, is the material used fo If YES, which one(s):	rt in an interlaboratory c eference material for this or the validation of proced or calibration of instrumer	omparison for thi s type of analysis ures?	is type of analysis c	Yes	No O	
12. 13.	Does your laboratory take par Yes No If yes, which one(s): Does your laboratory use a re Yes No If YES, is the material used fo If YES, is the material used fo If YES, which one(s): Do you have any comments?	rt in an interlaboratory c eference material for this or the validation of proced or calibration of instrumer	omparison for thi s type of analysis ures? nts?	is type of analysis c	Yes	No O	

Annex 6: Experimental details

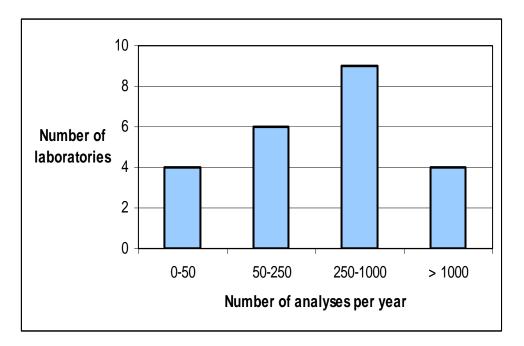
Lab Code	SOP?	Which SOP?	Sample pre- treatment	Digestion step	Extraction / separation	Instrument calibration
L01	No		Milling if necessary	Bomb digestion. 0.4 dry material is digested with 3 ml of conc. nitric acid for 3 h at 160 degrees in an oven	The acid residue is diluted to 10 ml with ultra-pure water (sample solution)	Calibration without standard addition Five standards 0, 0.05 0.10 , 0.25 and 0.50 µg/ml with the same acid conc as in the sample solution.
L02	No		None	Digestion with H ₂ O ₂ (30%), HF and HNO ₃ conc. by microwave high pressure (for Pb,Cd)	None	Add method; std solution Cd: 2 ppb; Pb: 50 ppb. Hg: square calibration with std solution 25-50- 100-150-300 ppb 1-2-5 ppm
L03	No		Add 2 ml 65% HNO ₃	microwave	dilution	4 step
L04	Yes	AOAC official method 999.10				
L05	No	EN 15550	No	HNO ₃ 25%	No	0 to 100 µg/l
L06	No		No pretreatment	Pb/Cd: Ashing at 470 C Hg: Wet digestion w. concentrated HNO ₃ /HCl/H ₂ O ₂	Pb/Cd: Extraction with 37% HCl and evaporation followed by extraction with 2% HCl. Hg: No extraction step	External calibration with correction for spike recovery.
L07	Yes					
L08	No	SR EN 13806/2003, SR EN 14082/2003, R (EC)18/2007, R (EC)1881/2006	We weigh approximately 0.5g of the sample and we add 10 ml of HNO ₃ 65% suprapur	We digest the sample in a microwave oven	After the digestion we dilute the sample up to 20 ml	we calibrate the instrument using standard solutions of Cd (0.2 -2.0 ng/ml), of Pb(5.0-50.0ng/ml)
L09	Yes					
L10	Yes					

Lab Code	SOP?	Which SOP?	Sample pre- treatment	Digestion step	Extraction / separation	Instrument calibration
L11	Yes	National Feed Codex				
L12	Yes					
L13	No		After shaking and mixing for the homogeneity, one weighs of about 0.5 g of sample was placed in PFA tube	Wet digestion; HNO ₃ + H ₂ O ₂ at 180°C	none	External calibration
L14	No	AOAC	Weigh, dreing, disolving	dry in oven, digestet in muffle furnace, disolved in 1N HCl, Hg - AMA 254	Extraction	Calibration curve for Pb (10-60 ug/l), calibration curve for Cd (1-10 ug/l), Hg (0.05-5ug/ml
L15	Yes	EU- Guideline; VdLUFA				
L16	Yes	ASU L00.00- 19/3				
L17	No	EN 14084 for Cd and Pb; EPA 7473 for Hg	Microwave	5 % NO ₃ H	No	ICP-MS
L18	No	ANSWER	None	Microwave digestion at 200°C for 25 min with HNO_3 and H_2O_2	None	ICP calibration with known solution. Calibration curves (5 concentrations) were used for each metal
L19	Yes					
L20	Yes	AOAC 999.10 (2005), AOAC 974.14 (2000), EN 13806 (2002)				
L21	Yes	NMKL				

Lab Code	SOP?	Which SOP?	Sample pre- treatment	Digestion step	Extraction / separation	Instrument calibration
L22	No	AOAC Official Method 999.10, AOAC Official Methods of Analysis, 2000, Chapter 9, p16- 19.	Dry ashing	dissolving the ash in diluted hydrochloric acid	Complexes of Pb and Cd with DDC are extracted into MIBK.	Matrix matched calibration curve
L23	No		None	0.5g sample with 5ml HNO ₃ microwave digestion at 200 celsius for15minutes	none	yes
L24	Yes	Cd, Pb - AOAC 999.11, Hg AOAC 971.21				
L25	Yes	ANSWER				
L26	Yes	ISO 17294- 2,2005				
L27	Yes	EN 14082				
L28	Yes	4 point calibration with similar matrix				
L29	No		None	microwave assisted digestion of 0,2 g dry sample with 1 ml H ₂ O, 3,5 ml HNO ₃ and 1 ml H ₂ O ₂ in the pressure bomb, made up to 10 g	None	linear through zero; 0,5 - 1- 2- 5 - 10 ppb in solution; corrected to added In-115 as internal standard
L30	No	SOP based on CEN method	we balanced 0,5 gr of the sample, we added 10 ml HNO ₃	microwave	No	No

Lab Code	SOP?	Which SOP?	Sample pre- treatment	Digestion step	Extraction / separation	Instrument calibration
L31	Yes	EN 15510:2007				

Annex 7: Number of samples analysed per year



European Commission

EUR 23236 EN – Joint Research Centre – Institute for Reference Materials and Measurements Title: Report of the third interlaboratory comparison organised by the Community Reference Laboratory for Heavy Metals in Feed and Food. Total Cd, Pb and Hg and extractable Cd and Pb in feed. Author(s): M.B. de la Calle, J. van de Kreeke, I. Verbist, S. Bynens, P. Taylor Luxembourg: Office for Official Publications of the European Communities 2008 – 45 pp. – 21 x 29.7 cm EUR – Scientific and Technical Research series – ISSN 1018-5593 ISBN 978-92-79-08275-7 DOI 10.2787/1332

Abstract

The Institute for Reference Materials and Measurements (IRMM) of the Joint Research Centre, a Directorate-General of the European Commission, operates the Community Reference Laboratory for Heavy Metals in Feed and Food (CRL-HM). One of its core tasks is to organise interlaboratory comparisons (ILCs) among appointed National Reference Laboratories (NRLs). This report presents the results of the third ILC of the CRL-HM which focused on the determination of total Cd, Pb and Hg and extractable Cd and Pb in feed according to Directive 2002/32/EC of the European Parliament and of the Council on undesirable substances in animal feed.

The test material used in this exercise was a commercial compound feed for fish provided by the Centro di Referenza Nazionale per la Sorveglianza e il Controllo degli Alimenti per Animali, Istituto Zooprofilattico del Piemonte. The material, naturally contaminated, was processed, bottled, labelled and dispatched by the Reference Materials Unit of the IRMM. The samples were dispatched on the second half of October 2007. Each participant received one bottle containing approximately 20 g of test material. Thirty-one participants from 25 countries registered to the exercise of which 31 submitted results for total Cd and total Pb, 28 for total Hg, 26 for extractable Cd and 24 for extractable Pb.

The assigned values (X_{ref}) were provided by IRMM using isotope dilution-inductively coupled plasmamass spectrometry (ID-ICP-MS). The analytical uncertainty of X_{ref} , u_{char} , was calculated according to the ISO Guide to the Expression of Uncertainty in Measurement (GUM). Homogeneity and stability studies were subcontracted to Bundesanstalt für Materialforschung und –Prüfung (BAM). The uncertainties of the respective assigned values, u_{ref} , were calculated combining the analytical uncertainty, u_{char} , with a contribution for the between-bottle homogeneity, u_{bb} , and for the short term stability of the test material, u_{sts} . Participants were invited to report the uncertainty on their measurements. This was done by 29 laboratories for total Cd, 23 for total Pb, 25 for total Hg, 18 for extractable Cd and 13 for extractable Pb.

Laboratory results were rated with z and zeta scores in accordance with ISO 13528. Standard deviation for proficiency assessment (also called target standard deviation) were calculated using the modified Horwitz equation and were 22 % of the assigned value for total Pb and Hg and for extractable Pb and 16 % of the assigned value for total and extractable Cd.

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